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Di-μ-sulfato-bis{[bis(3,5-dimethylpyrazol-1-yl)methane]copper(II)}

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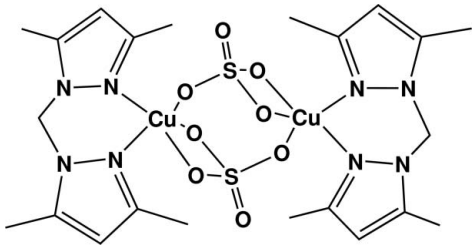
Key indicators: single-crystal X-ray study; T = 291 K; mean σ(C–C) = 0.006 Å; R factor = 0.046; wR factor = 0.113; data-to-parameter ratio = 13.3.

The molecule of the title compound, $[Cu_2(C_{11}H_{16}N_4)_2(SO_4)_2]$, sits on a center of symmetry. The Cu^{II} atom has a distorted trigonal–bipyramidal coordination geometry comprising three O atoms of the two symmetry-related SO_4^{2-} anions and two N atoms from one bis(3,5-dimethylpyrazol-1-yl)methane ligand.

Related literature

For related literature, see: Arnold *et al.* (2001); Dhar *et al.* (2004); Endres *et al.* (1984); Hatzidimitriou *et al.* (2006); He & Han (2006); Springsteen *et al.* (2006); Tamasi & Cini (2003); Thompson *et al.* (1998).



Experimental

Crystal data

$[Cu_2(SO_4)_2(C_{11}H_{16}N_4)_2]$
M_r = 727.76
Monoclinic, $P2_1/n$
a = 7.5293 (15) Å
b = 10.734 (2) Å
c = 17.740 (4) Å
β = 99.73 (3)°
V = 1413.2 (5) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 1.71 mm^{-1}
T = 291 (2) K
0.22 × 0.19 × 0.19 mm

Data collection

Rigaku Mercury diffractometer
Absorption correction: multi-scan (Jacobson, 1998)
T_{min} = 0.704, T_{max} = 0.737
13344 measured reflections
2580 independent reflections
2253 reflections with $I > 2\sigma(I)$
R_{int} = 0.039

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.045
$wR(F^2)$ = 0.112
S = 1.07
2580 reflections
194 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.46 e Å^{-3}
$\Delta\rho_{min}$ = −0.42 e Å^{-3}

Data collection: *CrystalClear* (Rigaku/MSC, 2001); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSC, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CS2085).

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supplementary materials

Acta Cryst. (2008). E**64**, m1242 [**doi:10.1107/S1600536808027840**]

Di-μ-sulfato-bis{[bis(3,5-dimethylpyrazol-1-yl)methane]copper(II)}

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Comment

SO_4^{2-} anion-bridged dimeric complexes of Cu(II) are reported extensively (Tamasi & Cini, 2003). In most of these structures the SO_4^{2-} anion acts as a bidentate bridge (Springsteen *et al.*, 2006; He & Han, 2006; Arnold *et al.*, 2001; Thompson *et al.*, 1998; Endres *et al.*, 1984). However, there are only two known examples of the tridentate bridge form (Hatzidimitriou *et al.*, 2006; Dhar *et al.*, 2004). The crystal structure of the title compound, $[Cu(bdmpm)(SO_4)]_2$ (*bdmpm* = bis(1,1-bis(3,5-dimethylpyrazol-1-yl)methane), shows a perfect centrosymmetric dimer, as two $\{Cu(bdmpm)\}^{2+}$ units are bridged by two sulfate anions in the complex (Fig. 1). The Cu···Cu distance is 3.769 (11) Å and the copper atom has a trigonal bipyramidal CuN_2O_3 coordination geometry with the sulfate O(2) atom and the N(1) atom as axial ligand atoms.

Experimental

The reaction of $CuSO_4.5H_2O$ (25 mg, 0.1 mmol) with *bdmpm* (22 mg, 0.11 mmol) in MeOH (10 ml) was carried out at ambient temperature for 10 minutes, the mixture was filtered and the filtrate was then left for crystallization.

Refinement

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å and U_{iso}(H) = 1.2 times U_{eq}(C).

Figures

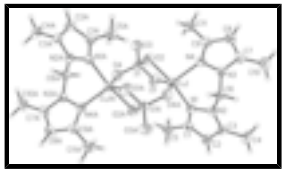

Fig. 1. The molecular structure with displacement ellipsoids drawn at the 50% probability level. Atoms labeled with the suffix A are related by the (-*x*, 1 - *y*, - *z*) symmetry operator.

Di-µ-sulfato-bis{[bis(3,5-dimethylpyrazol-1-yl)methane]copper(II)}

Crystal data

$[Cu_2(SO_4)_2(C_{11}H_{16}N_4)_2]$	F_{000} = 748
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Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation λ = 0.71073 Å
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a = 7.5293 (15) Å	θ = 3.0–25.4º

b = 10.734 (2) Å
c = 17.740 (4) Å
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V = 1413.2 (5) Å^3
Z = 2

μ = 1.71 mm^{-1}
T = 291 (2) K
Block, green
0.22 × 0.19 × 0.19 mm

Data collection

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Monochromator: graphite
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T = 291(2) K
ω scans
Absorption correction: multi-scan (Jacobson, 1998)
T_{min} = 0.704, T_{max} = 0.737
13344 measured reflections

2580 independent reflections
2253 reflections with $I > 2\sigma(I)$
R_{int} = 0.039
θ_{max} = 25.4º
θ_{min} = 3.0º
h = −9→9
k = −12→12
l = −21→21

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.045
$wR(F^2)$ = 0.112
S = 1.07
2580 reflections
194 parameters
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.054P)^2 + 2.0475P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.46 e Å^{-3}
$\Delta\rho_{min}$ = −0.42 e Å^{-3}
Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
Cu1	−0.00166 (6)	0.33244 (4)	0.03099 (2)	0.03196 (18)
S1	−0.08737 (12)	0.41689 (8)	−0.10738 (5)	0.0301 (2)
O1	−0.2040 (4)	0.3437 (3)	−0.06474 (16)	0.0422 (7)
O2	0.0920 (3)	0.4113 (3)	−0.05419 (15)	0.0431 (7)
O3	−0.0756 (4)	0.3645 (3)	−0.18102 (15)	0.0461 (7)
O4	−0.1457 (4)	0.5477 (3)	−0.11429 (16)	0.0444 (7)
N1	−0.1409 (4)	0.2717 (3)	0.10764 (17)	0.0316 (7)
N2	−0.0581 (4)	0.1991 (3)	0.16697 (17)	0.0330 (7)
N3	0.1848 (4)	0.1141 (3)	0.11191 (17)	0.0316 (7)
N4	0.1593 (4)	0.1722 (3)	0.04242 (17)	0.0315 (7)

C1	−0.3116 (5)	0.2839 (4)	0.1176 (2)	0.0343 (9)
C2	−0.3360 (6)	0.2205 (4)	0.1837 (2)	0.0411 (10)
H2	−0.4423	0.2152	0.2036	0.049*
C3	−0.1745 (6)	0.1674 (4)	0.2137 (2)	0.0365 (9)
C4	−0.1213 (7)	0.0889 (5)	0.2836 (3)	0.0557 (13)
H4A	−0.0814	0.0089	0.2691	0.084*
H4B	−0.2230	0.0785	0.3092	0.084*
H4C	−0.0253	0.1291	0.3175	0.084*
C5	−0.4425 (5)	0.3586 (4)	0.0638 (2)	0.0433 (10)
H5A	−0.3832	0.4307	0.0478	0.065*
H5B	−0.5399	0.3843	0.0888	0.065*
H5C	−0.4889	0.3089	0.0198	0.065*
C6	0.1344 (5)	0.1810 (4)	0.1760 (2)	0.0331 (8)
H6A	0.1751	0.1348	0.2228	0.040*
H6B	0.1939	0.2615	0.1807	0.040*
C7	0.2597 (5)	−0.0001 (4)	0.1088 (2)	0.0360 (9)
C8	0.2818 (6)	−0.0156 (4)	0.0345 (2)	0.0418 (10)
H8	0.3300	−0.0851	0.0140	0.050*
C9	0.2187 (5)	0.0921 (4)	−0.0045 (2)	0.0340 (9)
C10	0.3072 (7)	−0.0825 (4)	0.1768 (3)	0.0509 (11)
H10A	0.4025	−0.0449	0.2123	0.076*
H10B	0.3462	−0.1619	0.1609	0.076*
H10C	0.2034	−0.0936	0.2009	0.076*
C11	0.2141 (6)	0.1216 (5)	−0.0874 (2)	0.0495 (11)
H11A	0.0912	0.1305	−0.1123	0.074*
H11B	0.2701	0.0552	−0.1111	0.074*
H11C	0.2779	0.1979	−0.0920	0.074*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cu1	0.0298 (3)	0.0404 (3)	0.0263 (3)	0.00246 (19)	0.0063 (2)	0.00527 (19)
S1	0.0345 (5)	0.0300 (5)	0.0251 (5)	−0.0015 (4)	0.0034 (4)	−0.0014 (4)
O1	0.0415 (16)	0.0482 (17)	0.0366 (15)	−0.0127 (13)	0.0059 (13)	0.0060 (13)
O2	0.0328 (15)	0.064 (2)	0.0313 (14)	−0.0037 (13)	0.0025 (12)	0.0067 (14)
O3	0.063 (2)	0.0462 (17)	0.0285 (14)	0.0050 (14)	0.0053 (14)	−0.0078 (13)
O4	0.0621 (19)	0.0333 (15)	0.0372 (15)	0.0054 (14)	0.0068 (14)	0.0024 (12)
N1	0.0287 (16)	0.0381 (18)	0.0286 (16)	0.0030 (14)	0.0066 (13)	0.0039 (14)
N2	0.0355 (18)	0.0353 (17)	0.0287 (16)	0.0059 (14)	0.0069 (14)	0.0051 (14)
N3	0.0357 (17)	0.0299 (16)	0.0297 (16)	0.0044 (14)	0.0068 (14)	−0.0006 (14)
N4	0.0336 (17)	0.0320 (17)	0.0293 (16)	0.0045 (13)	0.0062 (14)	0.0010 (13)
C1	0.033 (2)	0.034 (2)	0.038 (2)	0.0010 (16)	0.0112 (17)	−0.0067 (17)
C2	0.041 (2)	0.044 (2)	0.043 (2)	−0.0049 (19)	0.0210 (19)	0.0026 (19)
C3	0.044 (2)	0.034 (2)	0.035 (2)	−0.0025 (17)	0.0155 (19)	0.0045 (17)
C4	0.068 (3)	0.059 (3)	0.044 (3)	0.004 (2)	0.023 (2)	0.023 (2)
C5	0.028 (2)	0.057 (3)	0.045 (2)	0.0065 (19)	0.0064 (19)	0.002 (2)
C6	0.039 (2)	0.035 (2)	0.0249 (18)	0.0016 (17)	0.0035 (16)	0.0009 (16)
C7	0.034 (2)	0.029 (2)	0.043 (2)	0.0028 (16)	0.0017 (17)	0.0007 (17)

C8	0.046 (2)	0.035 (2)	0.044 (2)	0.0067 (18)	0.0077 (19)	−0.0071 (19)
C9	0.031 (2)	0.034 (2)	0.037 (2)	0.0013 (16)	0.0057 (17)	−0.0069 (17)
C10	0.069 (3)	0.035 (2)	0.047 (3)	0.011 (2)	0.003 (2)	0.005 (2)
C11	0.057 (3)	0.059 (3)	0.034 (2)	0.010 (2)	0.011 (2)	−0.008 (2)

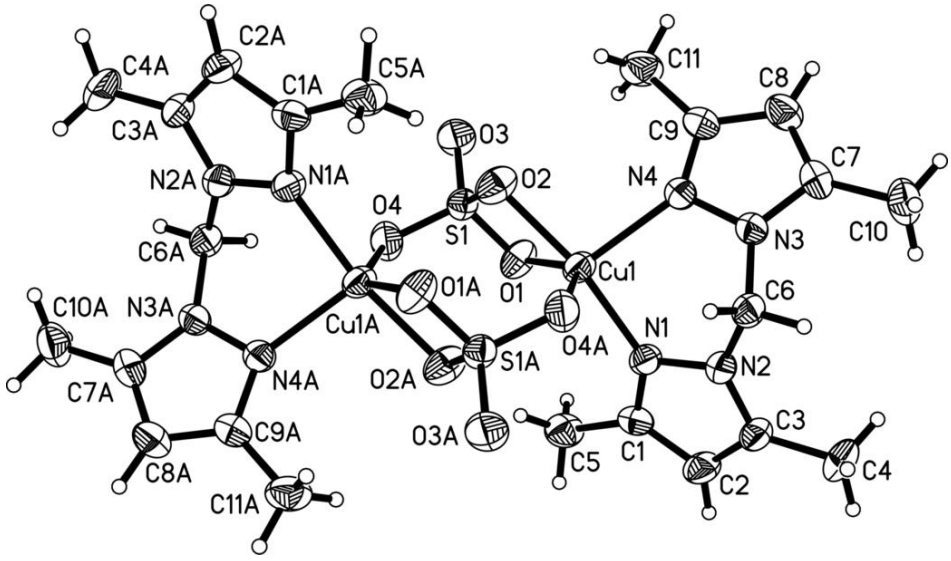
Geometric parameters (Å, °)

Cu1—N1	1.963 (3)	C2—H2	0.9300
Cu1—O2	1.964 (3)	C3—C4	1.497 (6)
Cu1—O1	2.085 (3)	C4—H4A	0.9600
Cu1—N4	2.094 (3)	C4—H4B	0.9600
Cu1—O4^{i}	2.125 (3)	C4—H4C	0.9600
Cu1—S1	2.5939 (11)	C5—H5A	0.9600
S1—O3	1.439 (3)	C5—H5B	0.9600
S1—O4	1.470 (3)	C5—H5C	0.9600
S1—O1	1.479 (3)	C6—H6A	0.9700
S1—O2	1.513 (3)	C6—H6B	0.9700
O4—Cu1^{i}	2.125 (3)	C7—C8	1.365 (5)
N1—C1	1.333 (5)	C7—C10	1.489 (6)
N1—N2	1.372 (4)	C8—C9	1.390 (6)
N2—C3	1.348 (5)	C8—H8	0.9300
N2—C6	1.444 (5)	C9—C11	1.499 (5)
N3—C7	1.354 (5)	C10—H10A	0.9600
N3—N4	1.366 (4)	C10—H10B	0.9600
N3—C6	1.449 (5)	C10—H10C	0.9600
N4—C9	1.326 (5)	C11—H11A	0.9600
C1—C2	1.394 (5)	C11—H11B	0.9600
C1—C5	1.486 (6)	C11—H11C	0.9600
C2—C3	1.367 (6)		
N1—Cu1—O2	168.29 (12)	C3—C2—H2	126.4
N1—Cu1—O1	100.44 (11)	C1—C2—H2	126.4
O2—Cu1—O1	69.90 (11)	N2—C3—C2	106.4 (3)
N1—Cu1—N4	91.60 (12)	N2—C3—C4	122.8 (4)
O2—Cu1—N4	98.70 (12)	C2—C3—C4	130.8 (4)
O1—Cu1—N4	117.24 (12)	C3—C4—H4A	109.5
N1—Cu1—O4^{i}	89.80 (12)	C3—C4—H4B	109.5
O2—Cu1—O4^{i}	93.50 (11)	H4A—C4—H4B	109.5
O1—Cu1—O4^{i}	139.06 (11)	C3—C4—H4C	109.5
N4—Cu1—O4^{i}	101.80 (12)	H4A—C4—H4C	109.5
N1—Cu1—S1	134.03 (9)	H4B—C4—H4C	109.5
O2—Cu1—S1	35.47 (8)	C1—C5—H5A	109.5
O1—Cu1—S1	34.73 (8)	C1—C5—H5B	109.5
N4—Cu1—S1	115.22 (9)	H5A—C5—H5B	109.5
O4^{i}—Cu1—S1	117.63 (8)	C1—C5—H5C	109.5
O3—S1—O4	111.17 (17)	H5A—C5—H5C	109.5
O3—S1—O1	112.93 (18)	H5B—C5—H5C	109.5
O4—S1—O1	110.82 (18)	N2—C6—N3	111.7 (3)

O3—S1—O2	111.43 (17)	N2—C6—H6A	109.3
O4—S1—O2	108.23 (17)	N3—C6—H6A	109.3
O1—S1—O2	101.81 (16)	N2—C6—H6B	109.3
O3—S1—Cu1	132.76 (13)	N3—C6—H6B	109.3
O4—S1—Cu1	115.83 (12)	H6A—C6—H6B	107.9
O1—S1—Cu1	53.44 (11)	N3—C7—C8	105.6 (3)
O2—S1—Cu1	48.89 (11)	N3—C7—C10	123.2 (4)
S1—O1—Cu1	91.83 (14)	C8—C7—C10	131.1 (4)
S1—O2—Cu1	95.64 (14)	C7—C8—C9	107.0 (4)
S1—O4—Cu1^{i}	114.00 (17)	C7—C8—H8	126.5
C1—N1—N2	106.1 (3)	C9—C8—H8	126.5
C1—N1—Cu1	134.3 (3)	N4—C9—C8	110.4 (3)
N2—N1—Cu1	119.6 (2)	N4—C9—C11	121.7 (4)
C3—N2—N1	110.9 (3)	C8—C9—C11	127.9 (4)
C3—N2—C6	129.9 (3)	C7—C10—H10A	109.5
N1—N2—C6	118.6 (3)	C7—C10—H10B	109.5
C7—N3—N4	111.8 (3)	H10A—C10—H10B	109.5
C7—N3—C6	130.3 (3)	C7—C10—H10C	109.5
N4—N3—C6	117.8 (3)	H10A—C10—H10C	109.5
C9—N4—N3	105.1 (3)	H10B—C10—H10C	109.5
C9—N4—Cu1	136.3 (3)	C9—C11—H11A	109.5
N3—N4—Cu1	117.0 (2)	C9—C11—H11B	109.5
N1—C1—C2	109.3 (3)	H11A—C11—H11B	109.5
N1—C1—C5	121.1 (3)	C9—C11—H11C	109.5
C2—C1—C5	129.6 (4)	H11A—C11—H11C	109.5
C3—C2—C1	107.3 (3)	H11B—C11—H11C	109.5

Symmetry codes: (i) $-x$, $-y+1$, $-z$.

Fig. 1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C1A
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C11
C8
C9
C7
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S1
N3
C10
C6A
O1A
O1
Cu1
C6
C10A
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C3
O3A
C9A
C5
C1
C4
C8A
C2
C11A